Name:	Date	Section			
Lab Partner: _(ONLINE)					
Data Sheet 1: Indirect (Back) Titration					
Directions:					

Use 1 tablet each for Tums & Equate. Measure out ~1.000g of Baking Soda. Record exact mass in mg.

Recorded Data: (All data should be recorded in pen.)

$$M_{acid} = M_{HCl} = 1.057$$
 $M_{base} = M_{NaOH} = 1.706$

Burette Readings (Record all measurements to the nearest 0.01 ml.)

		Baking Soda	Equate	TUMS
	Hydrochloric Acid, HCl	993 mg	(1 tablet)	(1 tablet)
1.	Initial Reading, V_i , ml	5.23	1.63	2.48
2.	Final Reading, V_f , ml	30.58	26.63	27.43
3.	Change in Volume, ml $\Delta V_{HCl} = V_f - V_i$			
4.*	$\begin{aligned} & \text{millimoles of HCl, } n_{\text{HCl}} \\ & n_{\text{HCl}} = \Delta V_{\text{HCl}} \text{ x } M_{\text{HCl}} \end{aligned}$			

	Sodium Hydroxide, NaOH	Baking Soda	Equate	TUMS
5.	Initial Reading,V _i , ml	27.05	2.30	16.58
6.	Final Reading, V _f , ml	34.85	16.58	27.05
7.	Change in Volume, ml $\Delta V_{NaOH} = V_f - V_i$			
8.*	$\begin{aligned} & \text{millimoles of NaOH, } n_{NaOH} \\ & n_{NaOH} = \Delta V_{NaOH} x M_{NaOH} \end{aligned}$			
9.*	$\begin{aligned} & \text{millimoles of excess HCl, } n_X \\ & n_X = n_{HCl} - n_{NaOH} \end{aligned}$			

^{*}On a separate sheet of paper, show calculations for the starred "*" items.

Name: D	Date	Section
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Lab Partner: _(ONLINE)____

		Baking Soda	Equ	ıate	TUMS
10.	Antacid Chemical Formula	NaHCO ₃	Al(OH) ₃	MgCO ₃	CaCO ₃
11.*	Molar Mass				
12.	mass of active ingredient (mg)	993	160	105	500
13.*	millimoles of active ingredient				
14.	Equivalents				
15.*	HCl theoretical (mmoles)				
16.	HCl actual (mmoles)				
17.*	Percent Yield (%)				
18.*	Theoretical mass of HCl used. (mg)				
19.*	Actual mass of HCl used. (mg)				
20.	Cost of Antacid per dose (\$)				
21.*	Theoretical Cost Effectiveness (\$/mg)				
22.	Actual Cost Effectiveness (\$/mg)				

For row #12 use the actual measured mass in mg of Baking Soda. For Tums & Equate, record the mass of the active ingredients in mg listed on the bottle.

Name:		_ Date	_ Section
Lab Partner:	(ONLINE)	_	

Data Sheet 2: Direct (Forward) Titration

Directions: Use 1 tablet for Tums. Measure out ~1.000g of Baking Soda. Record exact mass in mg. **Recorded Data:** (All data should be recorded in pen. Record all measurements to the nearest 0.01 ml.)

 $\mathbf{M}_{acid} = \mathbf{M}_{HCl} = \underline{\hspace{1cm}}$

		Baking Soda	TUMS
	Hydrochloric Acid, HCl	980 mg	(1 tablet)
23.	Initial Reading,V _i , ml	38.20	30.58
24.	Final Reading, V _f , ml	49.95	38.20
25.	Change in Volume, ml $\Delta V_{HCl} = V_f - V_i$		
26.	millimoles of HCl, n_{HCl} $n_{HCl} = \Delta V_{HCl} \times M_{HCl}$		
10.	Antacid Chemical Formula	NaHCO ₃	CaCO ₃
11.	Molar Mass		
12.	mass of active ingredient (mg)	980	500
13.	millimoles of active ingredient		
14.	Equivalents		
15.	HCl theoretical (mmoles)		
16.	HCl actual (mmoles)		
17.	Percent Yield (%)		

For row #12 use the actual measured mass in mg of Baking Soda. For Tums, record the mass of the active ingredients in mg listed on the bottle.

Nar	ne:	Date	Section
Lab	Partner: _(ONLINE)		
		Baking Soda	TUMS
18.	Theoretical mass of HCl used. (mg)		
19.	Actual mass of HCl used. (mg)		
20.	Cost of Antacid per dose (\$) (*See website below.)		
21.	Theoretical Cost Effectiveness (\$/mg)		
22.	Actual Cost Effectiveness (\$/mg)		
1. (Calculate Equivalents for Colu of HCl used are your mole e (You may have already done		
	NaHCO ₃ + HCl	\rightarrow	
	CaCO ₃ + HCl	\rightarrow	
	Al(OH) ₃ + HCl	\rightarrow	
	MgCO ₃ +HCl	\rightarrow	
	Mg(OH) ₂ + HCl	\vdash	

- **2.** Make 3 Column Graphs that include all 5 Trials (3 Indirect & 2 Indirect)
 - **Graph 1:** Graph the Antacid Effectiveness (Percent Yield or Row #17). [x-axis antacid name; y-axis percent yield]
 - **Graph 2:** Graph the Theoretical mg of HCl used vs. the Actual mg of HCl (Rows #18 & #19). [x-axis antacid name; y-axis Series 1 = Theoretical mg, Series 2 = Actual mg]
 - **Graph 3:** Graph the Theoretical vs. Actual Cost Effectiveness. (Rows #21 & 22) [x-axis antacid name; y-axis Series 1 Theoretical Cost, Series 2 Actual Cost]

^{*} http://web.mst.edu/~tbone/Subjects/TBone/antacidtable.html

Name:		Date	Section
Lab Partner: _(ONLINE)			
	Antacid Post La	ıb Questior	<u>1S</u>
Based on the Percer effective.	nt Yields (Graph 1) ra	ank the antaci	ds from most effective to least
2. Based on the Tums effective as the indirect titration	_	es in Graph 2	, were the direct titrations as
3. Based on Graph #3, theoretical cost effectiveness.	, rank the antacids fro	om least to mo	ost cost effective for the
4. Based on Graph #3, cost effectiveness.	, rank the antacids fro	om least to mo	ost cost effective for the actual
5. Which one did you	expect to be the mos	t cost effectiv	e? Were you correct? Explain.
	and 105 mg of MgCo could determine the t	O ₃ , instead of	the same price as the Equate and 160 mg of Al(OH) ₃ and 105 mg ectiveness of the new antacid
7. Aside from the cost someone might buy a given an			ss, give at least one other reason